

# **Iodocyclization of 1,4-Dihydropyridines**

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#### Abstract

Iodine (or related species) addition to 1,4-dihydropyridines, with properly attached substituents at the nitrogen atom, leads to the corresponding 3-iodotetrahydropyridinium ions, which undergo an internal nucleophilic attack to furnish regio- and stereoselectively iodobi(poly)heterocyclic ring systems in good yields. © 1998 Elsevier Science Ltd. All rights reserved.

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The chemistry of dihydropyridines shows a broad variety of synthetic transformations [1 - 6]. involving reductions, concerted reactions, Lewis or protic acid promoted additions, etc., which make these compounds valuable synthetic intermediates, especially in the fields of Natural Product Synthesis and Medicinal Chemistry [7 - 10]. However, the easy oxidation of 1,4dihydropyridines to the corresponding pyridinium salts (NADH is converted to NAD<sup>+</sup> in many metabolic reductions) seriously restricts their use in organic synthesis. Recently, we have accomplished "non-biomimetic" oxidations of N-alkyl-1,4-dihydropyridines, including formal epoxidations [11], diaminations [12], and alcoxyhalogenations [13], in simple experimental procedures that effectively avoid the natural oxidation route. Here we report the extension of this methodology to the preparation of polycyclic β-iodotetrahydropyridines through an intramolecular nucleophilic attack upon the initially generated iminium ion (Scheme 1).

Scheme 1

The starting 1,4-dihydropyridines 1 were prepared from the corresponding pyridinium salts by sodium dithionite reduction [14, 15]. Special care should be taken with the hydroxy substituted derivatives 1a-c because of their propensity to undergo cyclization [16].

The first experiments were performed from 1-(2-hydroxyethyl)-1,4-dihydropyridine 1a, which was treated with NIS in THF solution to stereoselectively afford the trans oxazolidine 2a

## (30 %, Scheme 2).<sup>1,2</sup>

The *anti* addition observed is well precedented in intramolecular (as well as intermolecular) additions to iminium ions [17], and was also confirmed by the <sup>1</sup>H-<sup>1</sup>H coupling constant analysis and COSY and ROESY experiments. Similarly, treatment of dihydropyridine 1b with iodine (1 equivalent) gave the iodinated oxazine 2b (22%) and compound 3 (33%). When the reaction was performed in the presence of excess NaHCO<sub>3</sub> (to scavenge the acid produced in the cyclization step) the latter product was not detected, and the yield of 2b<sup>3</sup> went up to 67%.

Chiral dihydropyridines have emerged as powerful tools in alkaloid synthesis [18 - 21]; to further expand the scope of the above iodocyclizations, we envisaged a similar process starting from the chiral non-racemic dihydropyridine 1c, which was prepared from the corresponding 1-(2,4-dinitrophenyl)pyridinium salt and (R)-(-)-2-phenylglycinol following a previously reported procedure [16]. This unstable compound was immediately allowed to react with NIS in THF

1. All new compounds gave satisfactory spectroscopic data ( $^{1}$ H and  $^{13}$ C NMR, IR, UV, MS) as well as elemental analysis or HRMS. 2. Oxazolidine **2a**.  $^{1}$ H NMR (CDCl<sub>3</sub>, 300 MHz) 7.02 (d, J = 1.2 Hz, 1H), 4.90 (d, J = 8.9 Hz), 4.16 (m, 1H), 3.99 (m, 1H), 3.83 (m, 1H), 3.66 (m, 1H), 3.47 (m, 1H), 2.86 (m, 2H);  $^{13}$ C NMR (CDCl<sub>3</sub>, 75.4 MHz) 142.9, 119.9, 90.5, 65.1, 62.6, 49.1, 33.8, 19.4; IR (KBr) 2189, 1615; UV (MeOH) 270 (4.0); MS (EI) 276 (M $^{+}$ , 59), 149 (100). HRMS Calcd for  $C_8H_9N_2OI$  275.9760, found 275.9760. 3. Oxazine **2b**.  $^{1}$ H NMR (CDCl<sub>3</sub>, 300 MHz) 7.23 (d, J = 1.5 Hz, 1H), 4.66 (d, J = 3.6 Hz, 1H), 4.29 (m, 1H), 4.16 (m, 1H), 3.84 (m, 1H), 3.70 (s, 3H), 3.49 (m, 2H), 2.98 (m, J = 17.5, 4.8, 1.5 Hz, 1H), 2.80 (dd, J = 17.5, 4.1 Hz, 1H), 1.97 (m, 1H), 1.49 (d, J = 13.4 Hz, 1H);  $^{13}$ C NMR (CDCl<sub>3</sub>, 75.4 MHz) 168.0, 142.5, 98.1, 87.4, 67.9, 51.7, 51.0, 28.0, 26.5, 20.7; IR (KBr) 1689, 1628; UV (MeOH) 279 (4.22); MS (EI) 323 (M $^{+}$ , 22), 292 (13), 196 (100). HRMS Calcd for  $C_{10}H_{14}NO_3I$  323.0018, found 323.0020.

solution to afford a diastereomeric mixture of iodotetrahydropyridines, which were subjected to ketal hydrolysis by stirring a CH<sub>2</sub>Cl<sub>2</sub> solution over silica gel, yielding 2c<sup>4</sup> with moderate diastereoselection (3: 1 ratio) over the minor isomer 2c<sup>3</sup> (overall yield, 38%).

The stereochemistry was determined by mono- and bidimensional NMR experiments (COSY, NOE, ROESY, HMQC), and follows trends similar to those previously described for the non-iodinated analogs [16] (see Figure 1).

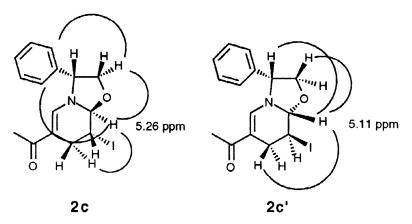


Figure 1. Selected NOE's for 2c and 2c'

Next, we explored the possibility of trapping the oxidatively formed iminium ion through electrophilic addition upon an aromatic ring. After several unsuccessful attempts to cyclize diversely substituted N-phenethyl-1,4-dihydropyridines, in which only intermolecular addition was observed [22],<sup>5</sup> the interaction of N-tryptophyl derivative 1d with NIS was studied and, to our delight, indoloquinolizidine 2d<sup>6</sup> was stereoselectively obtained in 85% yield (Scheme 2). This result opens interesting synthetic approaches towards indole alkaloids with the abovementioned framework (vincamine and tacamine types).

The interaction of an olefin moiety upon the iminium ion was then tested with homoallyl dihydropyridine 1e. Although treatment with iodine in the presence of methanol afforded the corresponding 2-methoxy-3-iodotetrahydropyridine, a THF solution of iodine cleanly yielded the somewhat unstable quinolizidine diiodide  $2e^7$  (43% isolated yield) as an epimeric mixture at

<sup>4.</sup> **2c.** <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 MHz) 7.41 - 7.33 (m, 5H), 7.22 (d, J = 1.5 Hz, 1H), 5.26 (d, J = 9 Hz, 1H), 4.69 (m, 1H), 4.53 (m, 1H), 3.89 (m, 1H), 3.81 (m, 1H), 3.43 (dd, J = 16.7, 5.2 Hz, 1H), 2.73 (m, J = 16.7, 12.5, 1.5 Hz, 1H), 2.06 (s, 3H); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 75.4 MHz) 192.7, 140.8, 136.9, 129.3, 128.9, 126.8, 111.8, 92.8, 73.7, 64.6, 32.0, 24.2, 21.9; UV (MeOH) 306 (4.20); MS (EI) 369 (M+, 30), 242 (100); [ $\alpha$ ]<sup>20</sup><sub>D</sub> -145.3 (c 0.6, MeOH); HRMS Calcd for C<sub>15</sub>H<sub>16</sub>NO<sub>2</sub>I 369.0226, found 369.0226.

<sup>5.</sup> For instance, treatment of methyl 1-phenethyl-1,4-dihydropyridine-3-carboxylate with iodine in McOH afforded the corresponding 3-iodo-2-methoxytetrahydropyridine, whereas when using NIS the corresponding 3-iodo-2-succinimide adduct was formed. The use of activated aromatic rings (4-methoxy and 3,4-dimethoxy substituted) and/or iodinating agents with less nucleophilic counterparts [bis(collidine)iodonium tetrafluoroborate] did not improve the situation, no benzoquinolizidines being detected from these experiments.

<sup>6.</sup> Indoloquinolizidine 2d. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 MHz) 8.80 (bs, 1II), 7.54 (s, 1H), 7.50 - 7.09 (m, 4H), 4.99 (d, J = 7.7 Hz, 1H), 4.57 (m, 1H), 3.74 (m, 1H), 3.68 (s, 3II), 3.50 (m, 1H), 3.12 (dd, J = 16.5, 5.0 Hz, 1H), 2.99 (dd, J = 16.5, 8.1 Hz, 1H), 2.97 - 2.75 (m, 2H); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 75.4 MHz) 168.5, 144.9, 135.8, 130.7, 126.4, 121.9, 119.3, 118.0, 111.2, 108.9, 96.6, 60.2, 51.4, 50.8, 30.0, 22.6, 21.8; IR (KBr) 3380, 1675, 1604; UV (MeOH) 292 (4.40), 221 (4.69); MS (EI) 408 (M<sup>+</sup>, 6), 280 (100), 221 (59); mp 137 - 138 °C (acetone - Et<sub>2</sub>O); correct combustion analysis.

<sup>7.</sup> Quinolizidine **2e** (major stereoisomer). <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 MHz) 7.28 (s, 1H), 4.16 (m, J = 7.1, 5.5 Hz, 1H), 3.68 (s, 3H), 3.40 (m, 1H), 3.36 - 2.31 (m, 9H), 2.12 (s, 3H); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 75.4 MHz) 167.6, 144.2, 95.8, 58.5, 51.5, 50.9, 50.2, 46.5, 41.3, 31.7, 31.5, 24.1; IR (NaCl) 1683, 1622; UV (MeOH) 289 (4.18); MS (EI) 461 (M<sup>+</sup>, 7), 334 (46), 206 (100); HRMS Calcd for  $C_{12}H_{17}NO_2I_2$  460.9348, found 460.9340.

the tetrasubstituted sp<sup>3</sup> carbon atom. This remarkable result suggests a stereocontrolled addition (anti) of the olefin moiety upon the iminium ion, followed by a non-stereospecific trapping of the resulting tertiary carbocation by iodide.

The methodology here disclosed is based on the "non-biomimetic" oxidation of dihydropyridines, and represents a new synthetic entry to a wide range of functionalized heterocyclic systems, potential precursors of bioactive or natural products.

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